

# The Calculation of the State Energies of *cis*- and *trans*-Terephthalaldehyde

By Kōzō INUZUKA

Department of Applied Science, Tokyo Electrical Engineering College, Chiyoda-ku, Tokyo

and Toshio YOKOTA

Tokyo University of Foreign Studies, Nishigahara, Kita-ku, Tokyo

(Received October 20, 1965)

Terephthalaldehyde is a carbonyl derivative of benzaldehyde and its *cis*- and *trans*-forms in solution have been found by Maki<sup>1)</sup> by means of ESR experiments.

Since *cis*-terephthalaldehyde has the  $C_{2v}$  symmetry, its  $\pi$ -electronic state belongs to the  $A_1$  or  $B_1$  of the  $C_{2v}$  symmetry group. Electronic transitions of  $A_1$  and  $B_1$  types have transition moments parallel to the  $z$  and  $x$  axes respectively.

On the other, *trans*-terephthalaldehyde has the  $C_{2h}$  symmetry and its  $\pi$ -electronic state belongs to the irreducible representation,  $A_g$  or  $B_u$ , of the symmetry group. Electronic transitions from the  $A_g$  ground state to the  $A_g$  excited states are impossible, but transitions to the  $B_u$  excited state are possible. Their transition moments are localized in the molecular plane.

We calculated the state energies for the two isomers by the semiempirical SCF-MO-CI method. The calculated procedure was the same as that used in a previous paper,<sup>2)</sup> in which the state energies of benzaldehyde were calculated. The molecular models of *cis*- and *trans*-terephthalaldehydes are shown in Fig. 1, in which the bond angle of  $C-C=O$  is assumed to be  $120^\circ$ .

## Results and Discussion

The SCF-MOs and MO energies are listed in Table I. In large molecules multiply-excited configurations should make a negligibly small contribution to the lower excitation energies. Therefore, we take account of single excitations only in the present calculation. The configuration

TABLE I. SCF-MOs AND THEIR ENERGIES OF *cis*-TEREPHTHALALDEHYDE (I) AND *trans*-TEREPHTHALALDEHYDE (II)

(I)	MO	Sym	$(\chi_1 \pm \chi_{10}^*)$	$(\chi_2 \pm \chi_9)$	$(\chi_3 \pm \chi_6)$	$(\chi_4 \pm \chi_5)$	$(\chi_7 \pm \chi_8)$	$\epsilon_i$ , eV.
	1	$b_2$	0.23522	0.22994	0.39085	0.36480	0.32551	-16.1377
	2	$a_2$	-0.53149	-0.39424	-0.22251	-0.08270	0.07577	-15.1986
	3	$b_2$	-0.53837	-0.32042	0.14416	0.21913	0.19670	-14.5752
	4	$b_2$	-0.00759	0.00637	0.03124	-0.48815	0.51053	-12.4567
	5	$a_2$	0.27285	0.03923	-0.53202	-0.27160	0.25923	-12.3713
	6	$b_2$	0.28375	-0.33686	-0.42022	0.24236	0.26587	-2.1367
	7	$a_2$	-0.06492	0.09752	0.02813	-0.53373	-0.44789	-1.0182
	8	$a_2$	-0.36143	0.54141	0.02278	-0.07749	0.26405	-0.4924
	9	$b_2$	0.27250	-0.48057	0.38587	-0.14792	-0.15500	0.6996
	10	$a_2$	0.09068	-0.20100	0.40760	-0.35849	0.39588	2.4959
(II)	MO	Sym	$(\chi_1 \pm \chi_{10}^{**})$	$(\chi_2 \pm \chi_9)$	$(\chi_3 \pm \chi_6)$	$(\chi_4 \pm \chi_7)$	$(\chi_5 \pm \chi_8)$	$\epsilon_i$ , eV.
	1	$a_u$	0.23700	0.23089	0.39093	0.35030	0.33910	-16.1380
	2	$b_g$	-0.53195	-0.39376	-0.22194	-0.08932	0.06883	-15.2072
	3	$a_u$	-0.53803	-0.31909	0.14595	0.21053	0.20755	-14.5834
	4	$b_g$	-0.12466	-0.02692	0.22092	0.56903	0.33338	-12.4904
	5	$b_g$	0.24247	0.02981	-0.48449	-0.02269	0.45285	-12.3486
	6	$a_u$	0.28233	-0.33625	-0.41978	0.29191	0.21401	-2.1527
	7	$a_u$	-0.02610	0.02992	0.02374	0.47582	-0.52102	-0.9905
	8	$b_g$	-0.36664	0.55020	0.02765	-0.17113	0.18110	-0.5149
	9	$a_u$	0.27196	-0.48054	0.38613	-0.14613	-0.15708	0.6953
	10	$b_g$	0.09091	-0.20152	0.40798	-0.37206	0.38243	0.4890

\* Double signs  $\pm$  mean  $\chi_1 + \chi_{10}$  for symmetry  $b_2$  and  $\chi_1 - \chi_{10}$  for  $a_2$ .

\*\*  $\chi_1 + \chi_{10}$  for symmetry  $a_u$  and  $\chi_1 - \chi_{10}$  for  $b_g$ .

1) A. H. Maki, *J. Chem. Phys.*, **35**, 761 (1961).

2) K. Inuzuka and T. Yokota, *This Bulletin*, **33**, 1055 (1965).

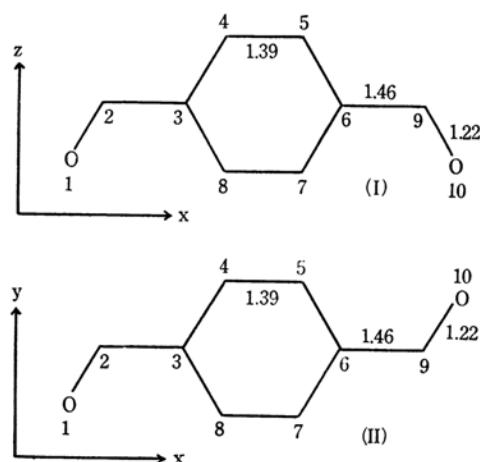


Fig. 1. Molecular models of *cis*- and *trans*-terephthalaldehydes (I and II).

mixing for *cis*-terephthalaldehyde leads to two 12th-order secular equations for the  $^1A_1$  and  $^3A_1$  states, and to two 13th-order secular equations for the  $^1B_1$  and  $^3B_1$  states. For *trans*-terephthalaldehyde, configuration mixing leads to two 12th-

TABLE II. CALCULATED ENERGY LEVELS OF TEREPHTHALALDEHYDE

G	<i>cis</i>			<i>trans</i>			Obs. eV.
	State	<i>E</i> , eV.	<i>f</i>	State	<i>E</i> , eV.	<i>f</i>	
	$^1A_1$	0		$^1A_g$	0		0
1	$^1A_1$	4.657	0.017	$^1B_u$	4.645	0.040	4.05
2	$^1B_1$	4.859	0.165	$^1B_u$	4.868	0.164	4.68
3	$^1B_1$	6.269	1.15	$^1B_u$	6.250	0.963	6.13
4	$^1A_1$	6.429	0.033	$^1A_g$	6.540	0	
5	$^1A_1$	6.823	0.704	$^1B_u$	6.736	0.962	
1	$^3B_1$	3.145		$^3B_u$	3.143		
2	$^3A_1$	3.845		$^3A_g$	3.856		
3	$^3B_1$	3.948		$^3B_u$	3.928		
4	$^3A_1$	4.029		$^3B_u$	4.042		

order secular equations for  $^1A_g$  and  $^3A_g$  states and to two 13th-order secular equations for  $^1B_u$  and  $^3B_u$  states. The state energies and wave functions are obtained from the solutions of these secular equations. The lower-lying state energies and oscillator strengths are listed in Table II.

We observed terephthalaldehyde to have three  $\pi-\pi^*$  absorption bands, at about 300, 260 and 200  $m\mu$ , in an *n*-hexane solution. The energy values corresponding to these absorption bands are also listed in Table II. The calculated values are in agreement with the observed values. From Table II we can not find any significant difference in state energies between the *cis*-isomer and the *trans*-isomer, but the 1st excited energy of the *trans*-isomer is lower than that of the *cis*-isomer. The values of the Coulomb integral are almost invariant over a great atomic distance. The change in the position of the carbonyl group from the *trans*-form to the *cis*-form does not have any appreciable effect on the molecular integral, because the distance between carbonyl groups is large. The similarity in electronic structure of *cis*- and *trans*-terephthalaldehydes is attributed to the above fact. The absorption spectrum of terephthalaldehyde resembles that of benzaldehyde. The spectrum of the former shifts to a longer wavelength side than that of the latter. The similarity in absorption spectra can be seen in the transition energies and oscillator strengths obtained for benzaldehyde<sup>2,3</sup> and for terephthalaldehyde. The charge densities and the free valences of both isomers for the ground and lower excited states are listed in Table III. These values of the excited states are calculated by assuming the same molecular structures as in the ground state. The charge densities of the position 4 of *cis*- and *trans*-terephthalaldehydes indicate a high reactivity to electrophilic reagents.

The present study was partially financed by the Scientific Research Fund of the Ministry of Education.

TABLE III. CHARGE DENSITIES ( $q_i$ ) AND FREE VALENCES ( $f_i$ ) OF *cis*-TEREPHTHALALDEHYDE (I) AND *trans*-TEREPHTHALALDEHYDE (II) IN THE GROUND AND LOWER EXCITED STATES

(I)	State										
		$q_1$	$q_2$	$q_3$	$q_4$	$q_7$	$f_2$	$f_3$	$f_4$	$f_7$	
	$^1A_1$	G	1.404	0.625	1.101	1.000	0.957	0.617	0.214	0.411	0.415
	$^1A_1$	1	1.431	0.707	1.036	0.951	0.875	0.605	0.360	0.710	0.711
	$^1B_1$	2	1.405	0.716	0.939	0.997	0.943	0.598	0.463	0.627	0.618
	$^1B_1$	3	1.381	0.684	0.989	1.019	0.926	0.677	0.339	0.759	0.733
(II)	State										
		$q_1$	$q_2$	$q_3$	$q_4$	$q_7$	$f_2$	$f_3$	$f_4$	$f_7$	
	$^1A_g$	G	1.406	0.624	1.014	0.999	0.958	0.618	0.215	0.412	0.413
	$^1B_u$	1	1.420	0.705	0.991	0.969	0.915	0.601	0.407	0.677	0.674
	$^1B_u$	2	1.415	0.706	0.979	0.995	0.905	0.599	0.418	0.649	0.678
	$^1B_u$	3	1.372	0.676	0.978	0.960	1.014	0.680	0.347	0.768	0.707

3) K. Inuzuka and T. Yokota, *ibid.*, **37**, 1224 (1964).